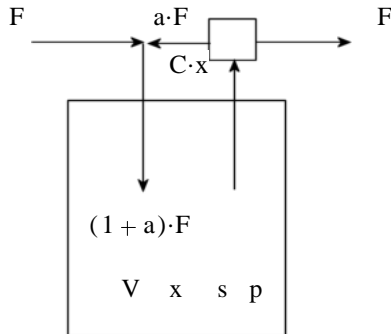


Find the steady-state values for a continuous bioreactor with **cell recycle**.

Instructor: Nam Sun Wang



F ... flow rate

a ... recycle ratio (Note that textbook uses the symbol  $\alpha$ , which denotes growth related product formation here.)

C ... biomass concentration factor

V ... bioreactor volume

x, s, p ... biomass, substrate, & product concentration

The kinetics of formation of the desired product is mixed (i.e., partially cell growth related and partially non-growth related). A cell-free feed stream continuously enters the fermentor of volume V at a flow rate of F, and the fermentor content is also continuously withdrawn. A cell separator concentrates the biomass by a factor of C, and the recycle loop has a flow rate of  $f=a \cdot F$ . Substrate and product are not concentrated. Assume that the limiting substrate is utilized only for cell growth but not for product formation. The following Contois equation of growth describes high-density cell fermentation where cell growth is suppressed when the fermentor becomes too crowded.

$$\mu(x, s) = \frac{\mu_m \cdot s}{K \cdot x + s} \quad \dots \text{Contois equation of cell growth}$$

### Analytical Solution

Steady-state conditions:

$$dx/dt = 0 = (\mu - (1 + a - a \cdot C) \cdot D) \cdot x \quad \dots \text{Biomass}$$

$$ds/dt = 0 = D \cdot (s_f - s) - \frac{1}{Y} \cdot \mu \cdot x \quad \dots \text{Substrate}$$

Given

$$dp/dt = 0 = \alpha \cdot \mu \cdot x + \beta \cdot x - D \cdot p \quad \dots \text{Product}$$

The non-washout solution of the above equations satisfies:

$$\mu = \frac{\mu_m \cdot s}{K \cdot x + s} \quad \mu = (1 + a - a \cdot C) \cdot D \quad x = \frac{Y \cdot (s_f - s)}{1 + a - a \cdot C}$$

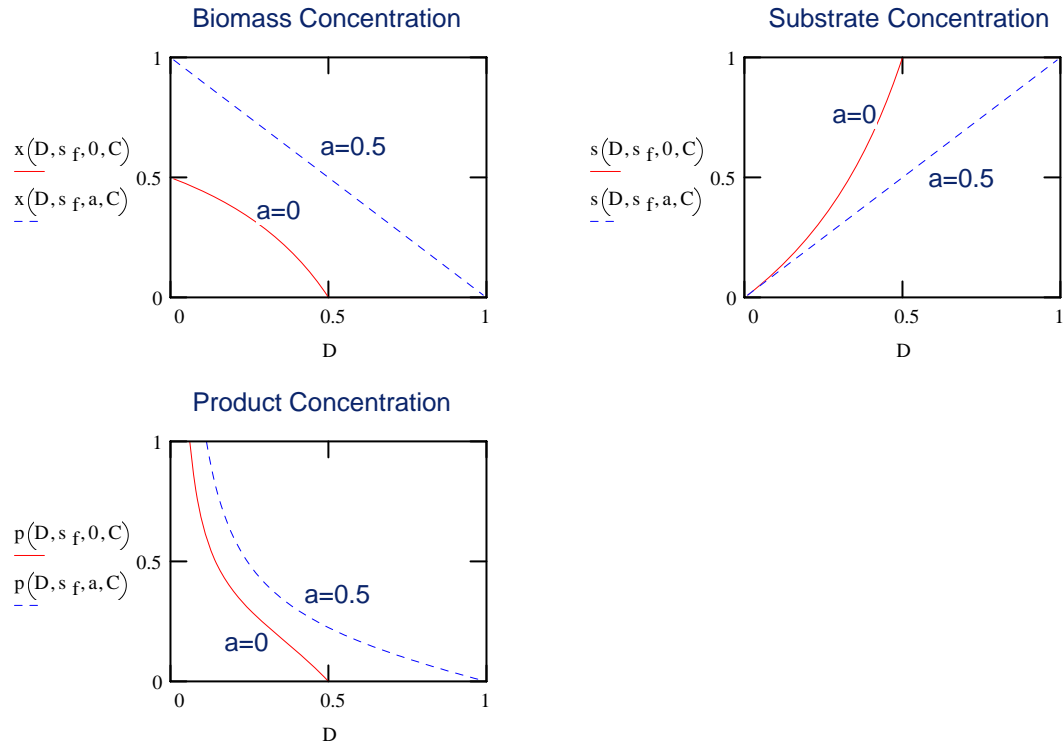
Find(x, s, p,  $\mu$ )  $\Rightarrow$

$$\left[ \begin{array}{l} -Y \cdot s_f \frac{(D \cdot a \cdot C - D \cdot a + \mu_m - D)}{(-K \cdot D \cdot Y - K \cdot D \cdot a \cdot Y + K \cdot D \cdot a \cdot C \cdot Y + D + 2 \cdot D \cdot a - 2 \cdot D \cdot a \cdot C + D \cdot a^2 - 2 \cdot D \cdot a^2 \cdot C + D \cdot a^2 \cdot C^2 - \mu_m)} \\ D \cdot K \cdot s_f \frac{Y}{(D \cdot a \cdot C - D \cdot a + K \cdot D \cdot Y + \mu_m - D)} \\ Y \cdot s_f \frac{(D \cdot a \cdot C - D \cdot a + \mu_m - D)}{(-K \cdot D \cdot Y - K \cdot D \cdot a \cdot Y + K \cdot D \cdot a \cdot C \cdot Y + D + 2 \cdot D \cdot a - 2 \cdot D \cdot a \cdot C + D \cdot a^2 - 2 \cdot D \cdot a^2 \cdot C + D \cdot a^2 \cdot C^2 - \mu_m - \mu_m \cdot a + \mu_m)} \\ D + D \cdot a - D \cdot a \cdot C \end{array} \right]$$

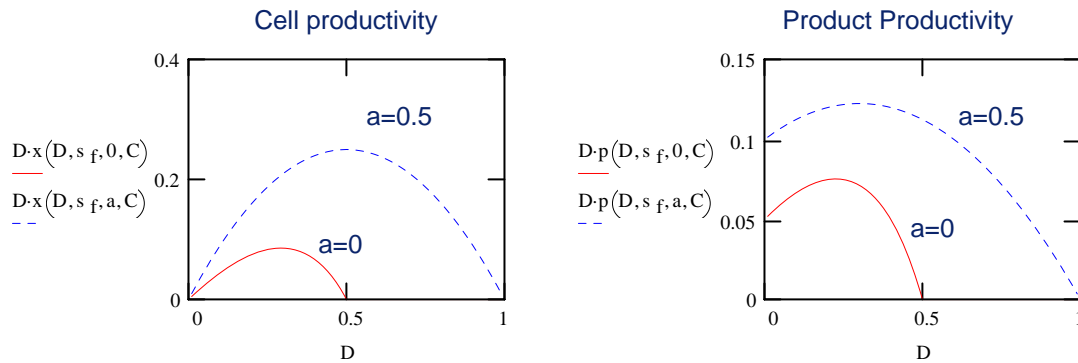


Comparison of steady-state variables /w and /wo cell recycle

$D := 0.01, 0.02 \dots 0.99$



Comparison of steady-state cell/product productivities /w and /wo cell recycle



It is clear from the graph that both cell and product concentrations are higher with cell recycle. Intuitively, cell recycle always increase the level of cells in a bioreactor. In turn, more cells yield more product. The same can be said of cell and product productivities.

$$\left. \begin{array}{l} \overline{(\mu_m - \mu_m \cdot a + \mu_m \cdot a \cdot C)} \\ \frac{(-D \cdot \alpha - \alpha \cdot D \cdot a + \alpha \cdot D \cdot a \cdot C - \beta)}{D} \cdot \frac{1}{m \cdot a \cdot C} \end{array} \right\}$$